



CATRIN

Czech Advanced
Technology and Research
Institute

Unleashing the Power: Superior Properties of Fluorographene-Derived Materials for Energy Storage Applications

Michal Otyepka



The hidden
gems of Europe ...



Olomouc and the Olomouc Region in Europe





Palacký University
Olomouc

Est. 1573
8 faculties
1 research unit (CATRIN)



Est. 2021
(history from 2010)



CATRIN-RCPTM
Nanotechnologies,
Chemistry

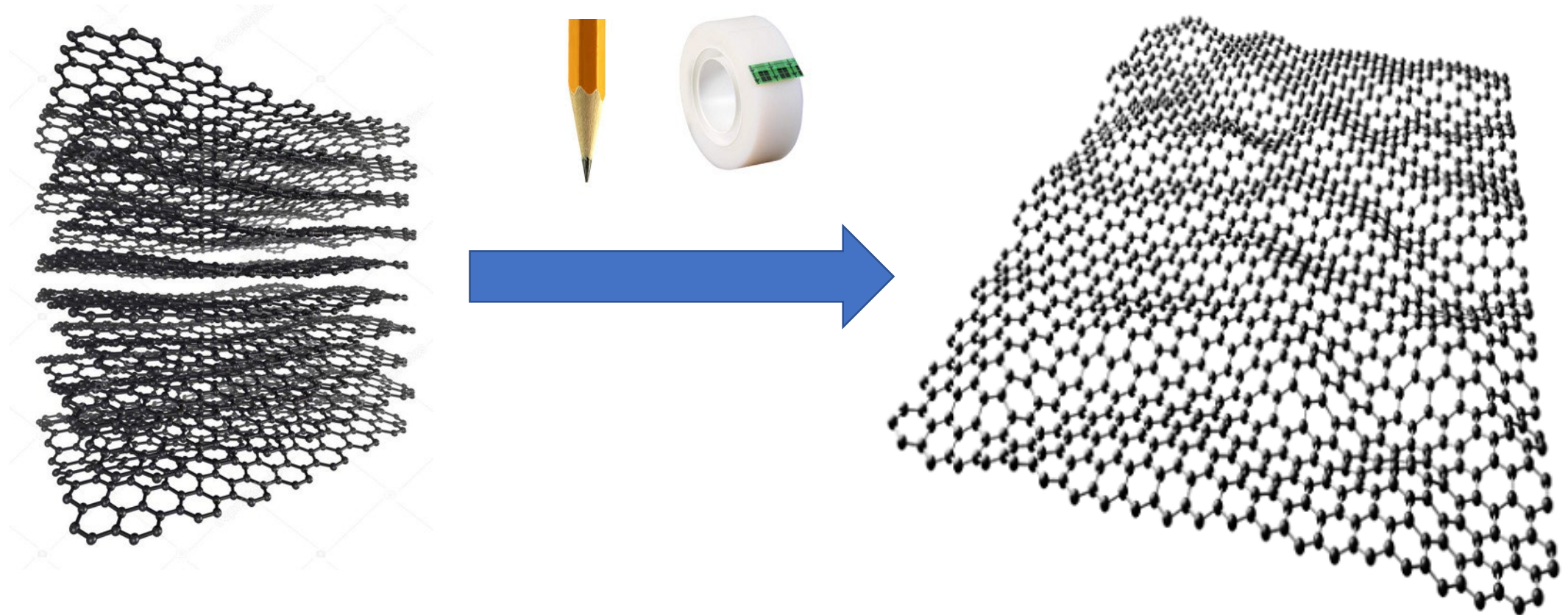


CATRIN-CRH
Biotechnologies, Agriculture



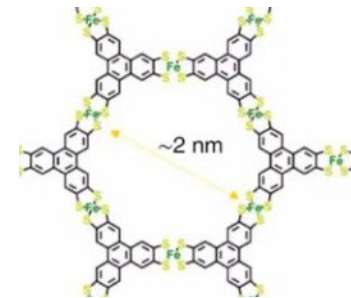
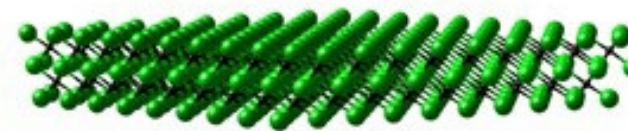
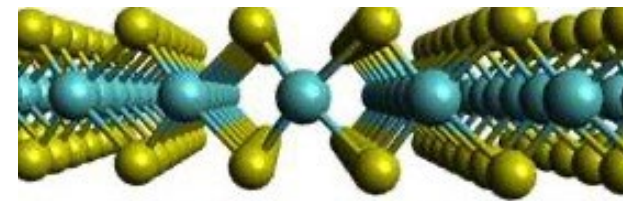
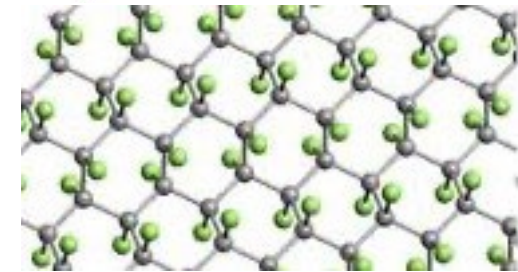
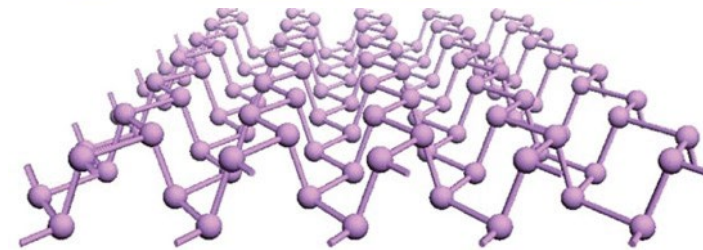
CATRIN-IMTM
Biomedicine

Graphene is 2D carbon material prepared by graphite exfoliation (Geim, Novoselov 2004)



2D Materials

- Graphene - 2004 by Novoselov and Geim
- Wide family of 2D materials
 - One element
 - C: graphene, P: phosphorene, ...
 - More elements
 - graphene derivatives – graphane (C_xH_x), fluorographene (C_xF_x), graphene oxide
 - graphene analogs – hBN
 - G- C_3N_4
 - MXenes (Ti_3C_2 ...)
 - transition metal chalcogenides (MoS_2 ...)
 - transition metal oxides and hydroxides (TiO_2 , ... $Ni(OH)_2$)
 - 2D zeolites
 - 2D MOFs, COFs



Fluorographene (discovered in 2010)

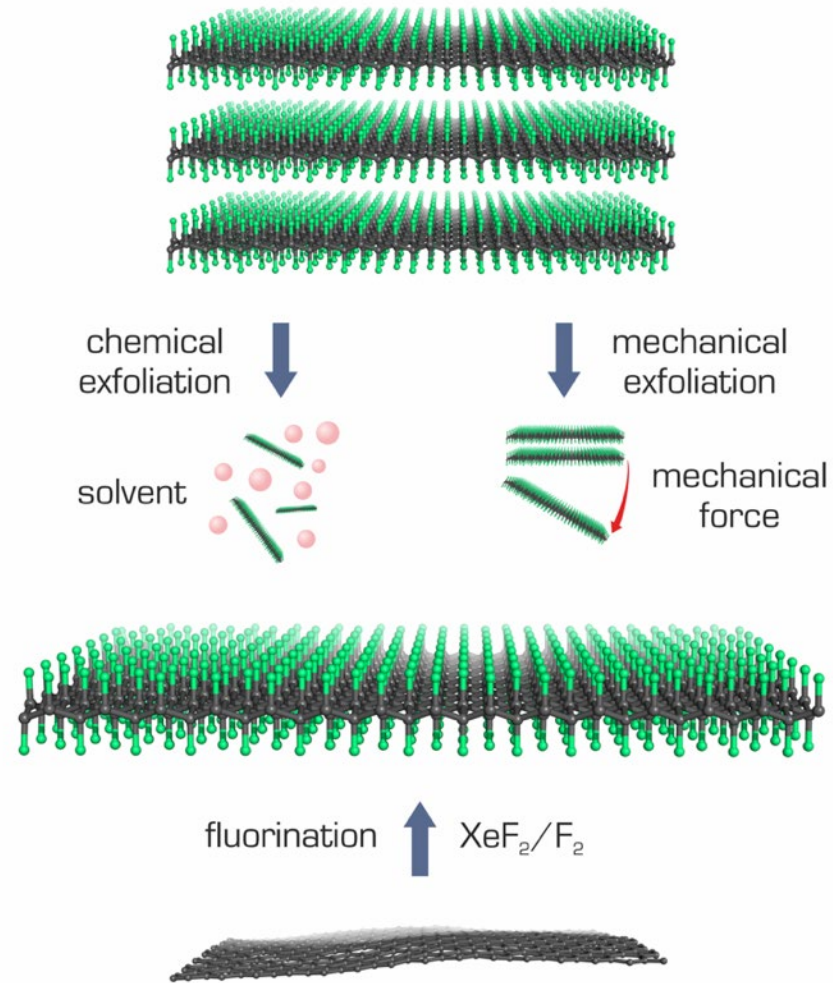
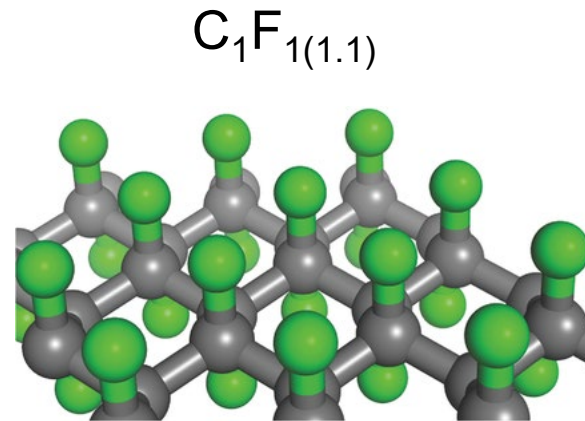
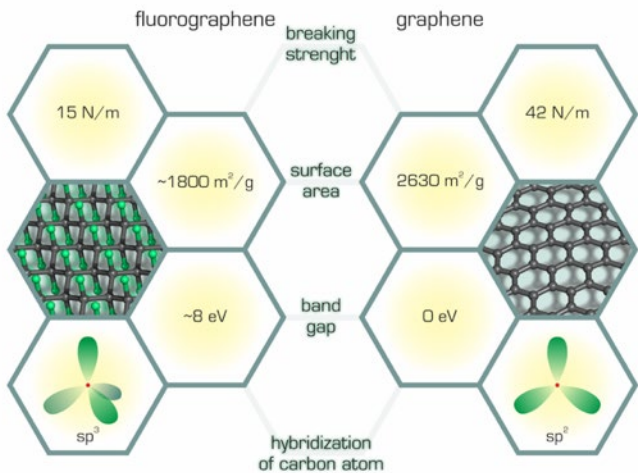
Mechanical exfoliation of graphite fluoride
Nair RR *et al.* Small 6, 2877 (2010)

Chemical exfoliation of graphite fluoride
Zbořil R *et al.* Small 6, 2885 (2010)

Fluorination of graphene

Robinson JT *et al.* Nano Letters 10, 3001 (2010)

Cheng SH *et al.* PRB 81, 205435 (2010)



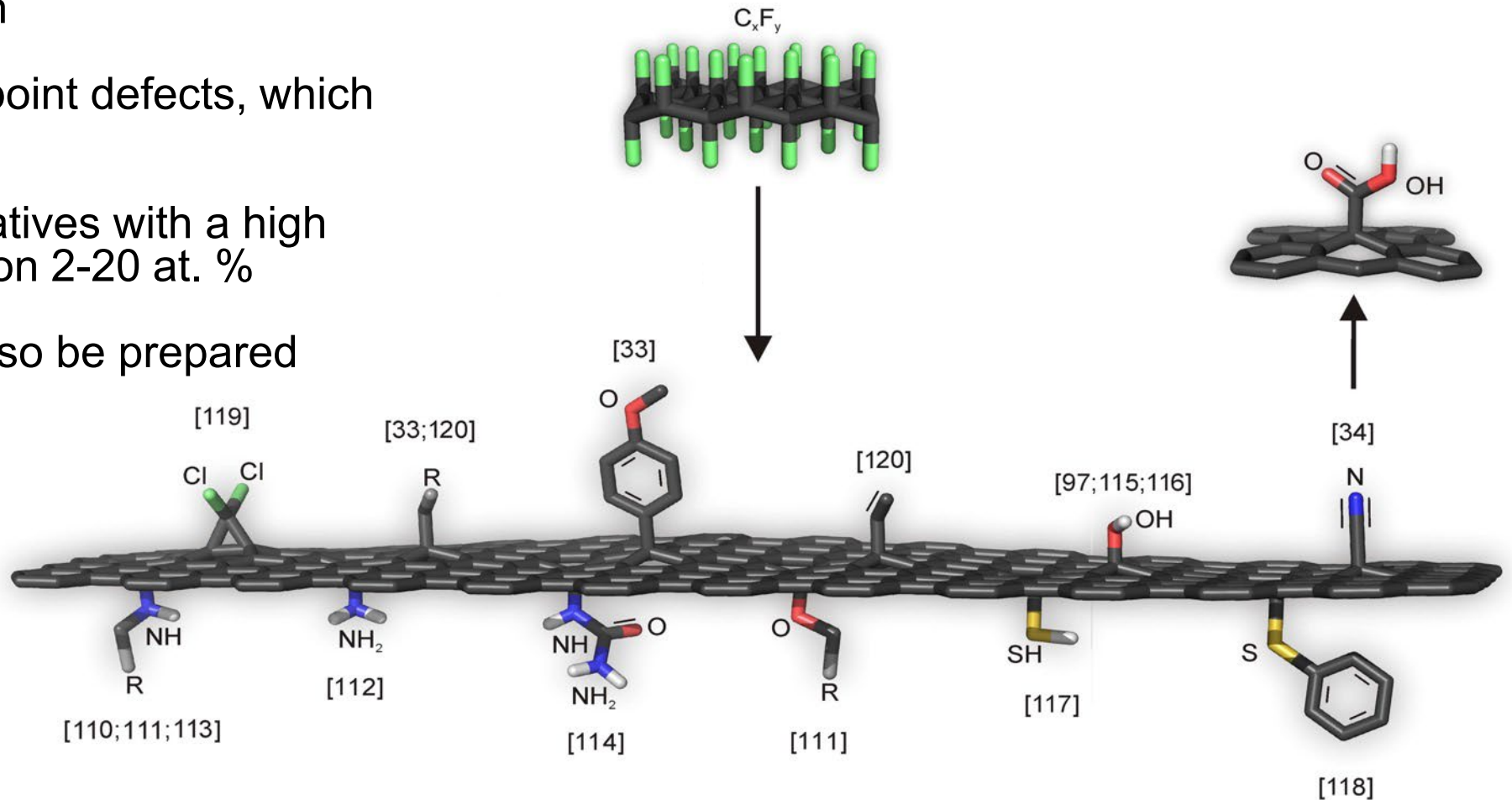
Fluorographene reacts at mild conditions

elimination + substitution

reaction is triggered by point defects, which behave like el-philes

leads to graphene derivatives with a high degree of functionalization 2-20 at. %

doped graphenes can also be prepared



Nanoscale 10, 4696, 2018

J. Phys. Chem. Lett. 9, 3580, 2018

ACS Sustainable Chem. Eng., 8, 4764, 2020

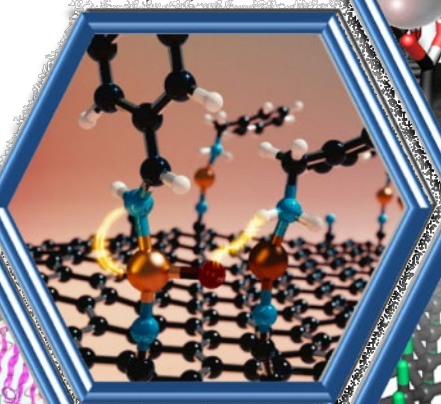
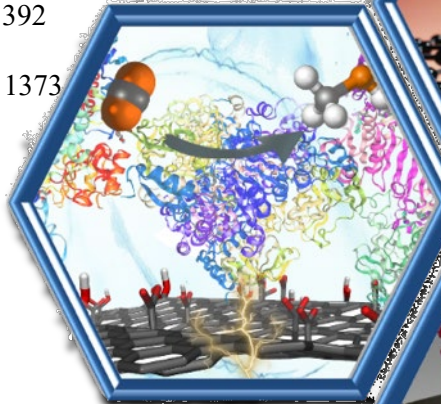
Appl. Mater. Today, 9, 60, 2017

[refs therein]

Applications explored in our lab for graphene-derivatives

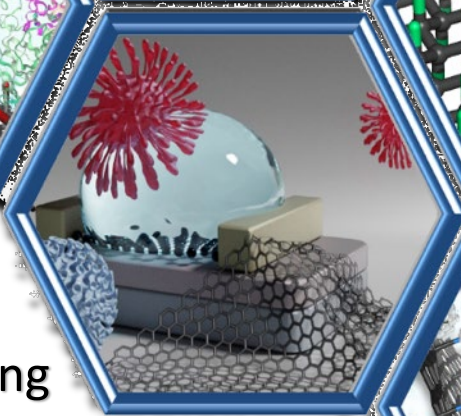
Adv. Mater. **2019**, 31, 1900323
Green Chemistry **2019**, 21, 5238
Chem. Sci. **2019**, 10, 9438
ACS Appl. Mater. Interfaces **2020**, 12, 250
Adv. Mater. Int. **2021**, 2001392
Small **2021**, 17, 2006477
Nature Commun. **2023**, 14, 1373

Catalysis



Biosens. Bioelectron. **2020**, 166, 112436
ACS Omega **2019**, 4, 19944
Biosens. Bioelectron. **2017**, 89, 532
Biosens. Bioelectron. **2021**, 195, 113628
Green Chem. **2023**, 25, 1647
Small **2023**, in press

Sensing

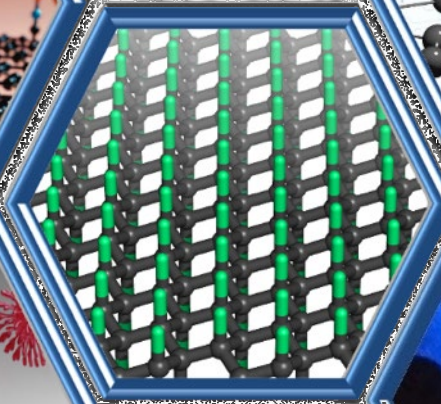
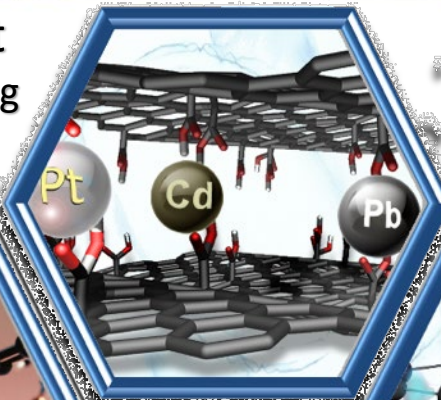


Antibacterial mat.; *Adv. Sci.* **2021**, 2003090

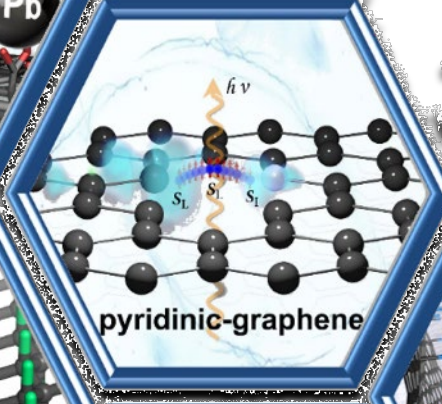
Nano-bio interface

Environment Detox-monitoring

ACS Nano **2021**, 15, 3349
Small **2022**, 18, 2201003



Spin control Magnetism

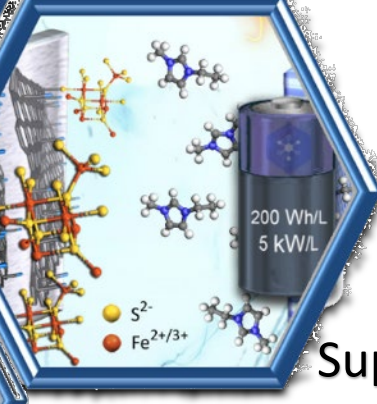


Batteries

Adv. Funct. Mater. **2021**, 2101326
Adv. Energy Mater. **2022**, 12, 2103010



Nat. Commun. **2017**, 8, 1
ACS Nano **2018**, 12, 12847
Nat. Commun. **2018**, 9, 1
Adv. Mater. **2019**, 31, 1902587
ACS Appl. Mater. Interfaces **2020** 12, 34074, 2020



Supercaps

Adv. Mater. **2018**, 30, 1705789
Adv. Funct. Mater. **2018**, 28, 1801111
Adv. Fun. Mater. **2019**, 27, 1906998
Chem. Mater. **2019**, 31, 4698
J. Mater. Chem. A **2020**, 8, 25716
Adv. Mater. **2021**, 33, 2004560
Env. En. Sci. **2022**, 15, 740



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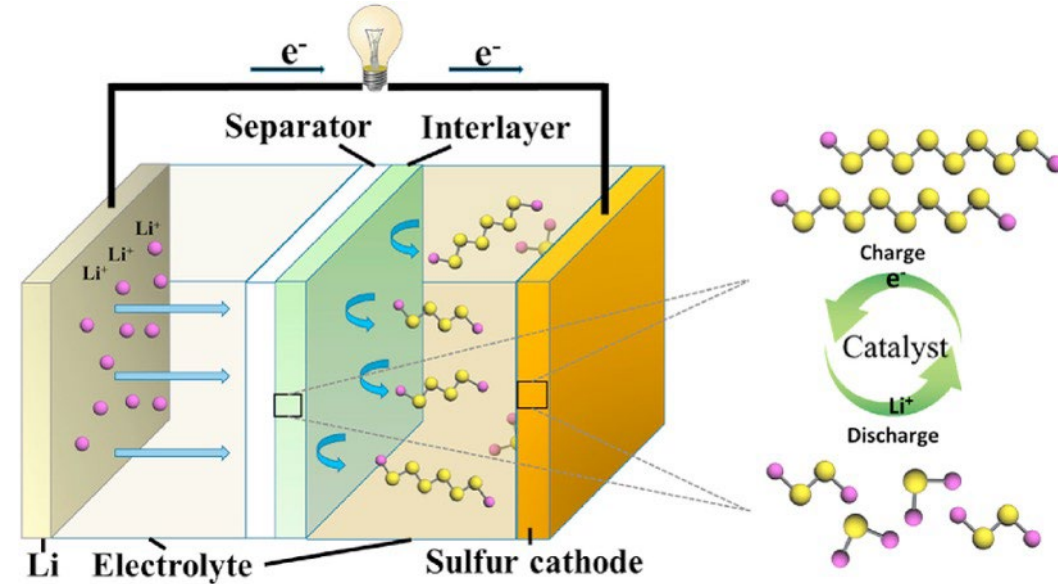
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Sulphurized Graphene as Cathode for LSB

... just a short detour

Lithium-Sulfur Batteries (LSB)

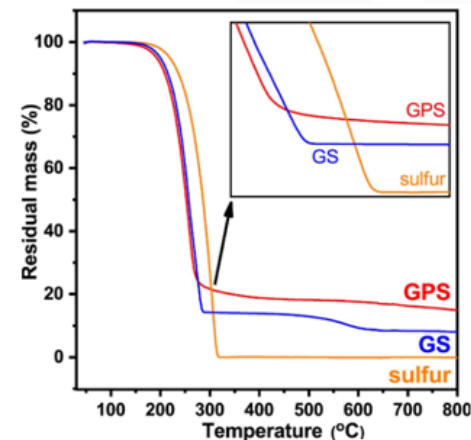
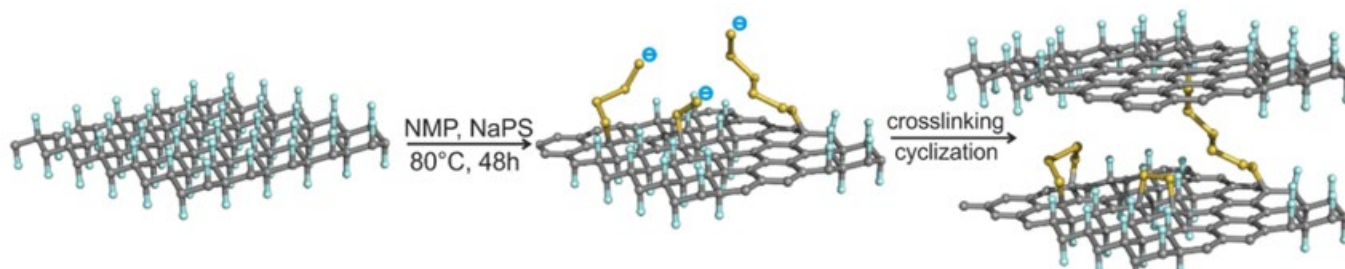
- a promising alternative for energy storage
- high theoretical capacity (1672 mAh g^{-1}) and specific energy (2600 Wh kg^{-1})
- sulfur is environmentally friendly and a key byproduct of the petroleum industry
- several bottlenecks hamper the practical development of the LSBs
 - sulfur's poor conductivity
 - large volume change
 - “shuttling effect” of lithium polysulfides (PSs), formed during the charge/discharge process. The dissolution of Li-PSs into the liquid electrolyte leads to low Coulombic efficiency, poor sulfur utilization, fast capacity fading, and other parasitic reactions with the Li anode.



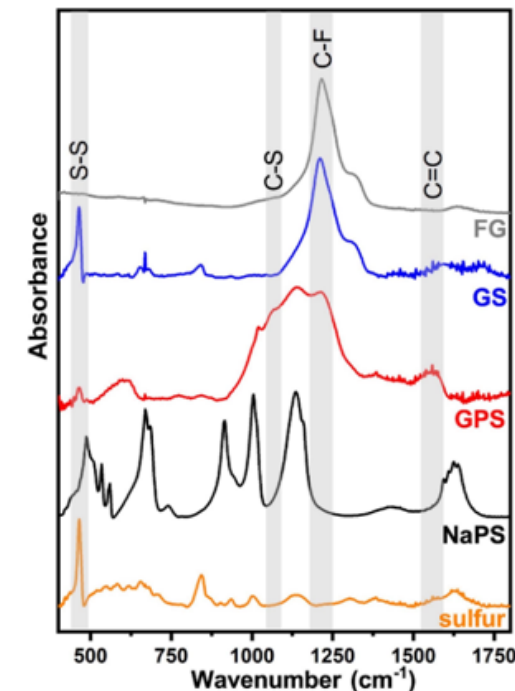
From: Energy Storage Materials 20, 55-70, 2019

Background and characterization

- highly and covalently sulfurized graphene cathode
- exploiting the nucleophilicity of polysulfide anions and the electrophilic centers in fluorographene
- Sulfur chains are immobilized by covalent bonding to graphene



Sulfur loading ~80 wt. %

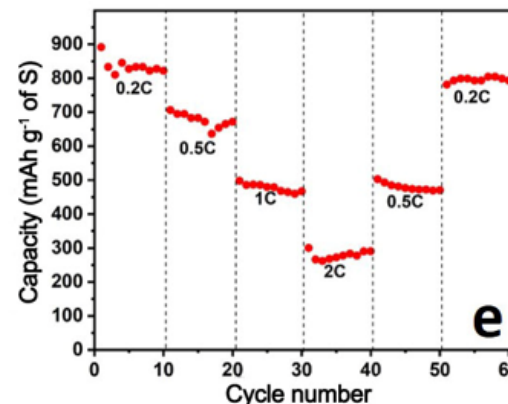
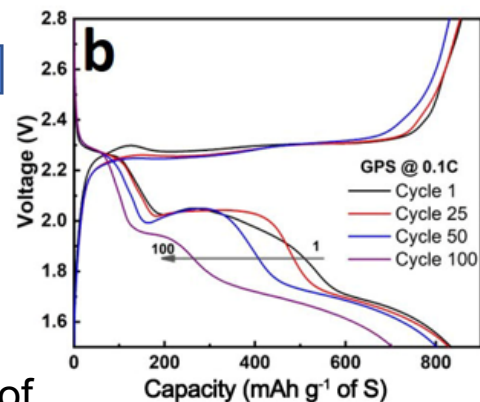


GPS: C–F at $\sim 1200\text{ cm}^{-1}$ decreased (i.e., defluorination) 1580 cm^{-1} band emerged (graphene lattice formation).

The new band at $\sim 1150\text{ cm}^{-1}$ demonstrates the development of covalent C–S bonds.

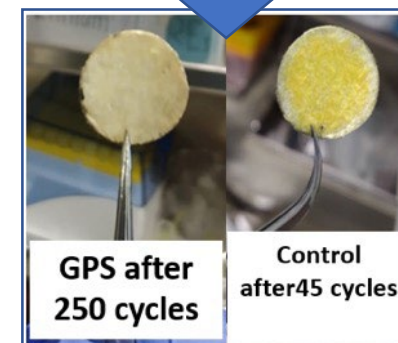
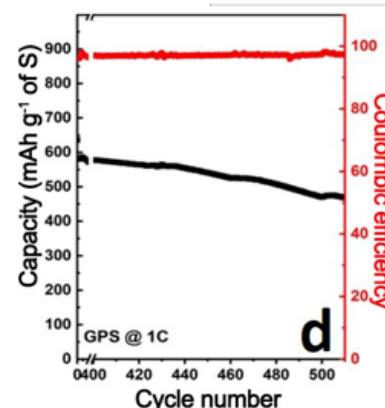
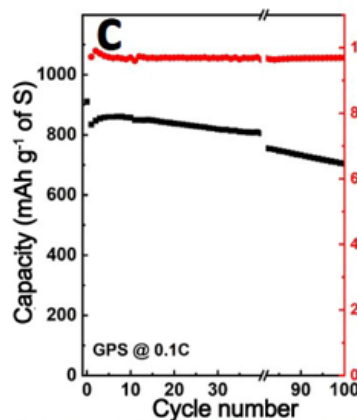
Electrochemical performance of the graphene-polysulfide cathode

High electrochemical reversibility for more than 50 cycles at 0.1 C (167 mA g^{-1})



Good rate capability for both high and low specific currents

Outstanding stability for high and low specific currents
reduced shuttling-effect



Glass fiber separator after cycling showing the dissolution of sulfur at the control material

- ✓ Alkylhalide-like and elegant chemistry of fluorinated carbon matrices exploitation
- ✓ Effective pathway for the development and study of previously unexplored cathode materials for LSBs.
- ✓ Electrochemical cycling of the sulfurized-graphene material against lithium exhibited top-rated performance with only **5 wt. %** of conductive additives and at low temperature of **25 °C**



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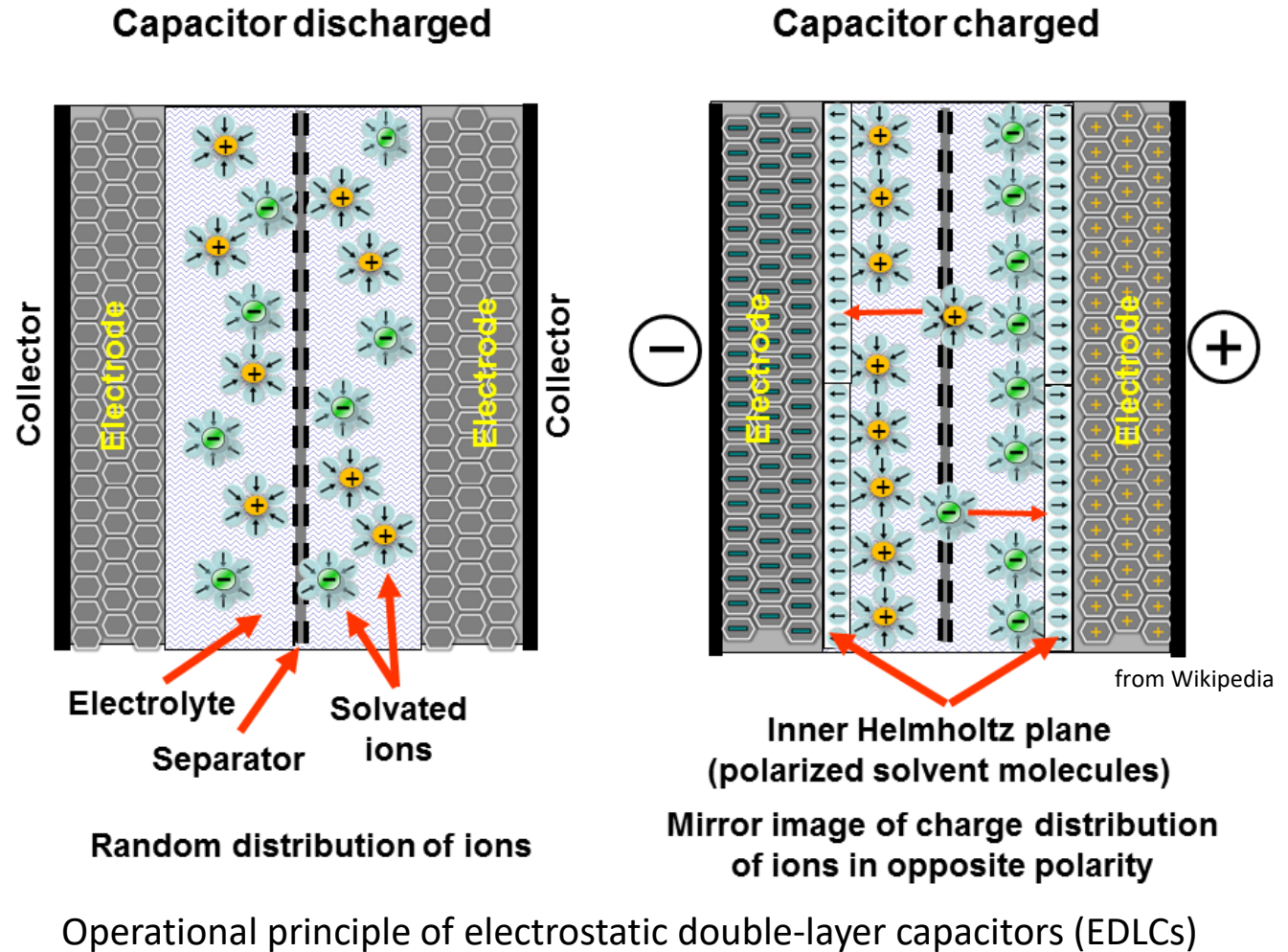
N-doped Graphene as A Supercapacitor Electrode Material

Supercapacitor

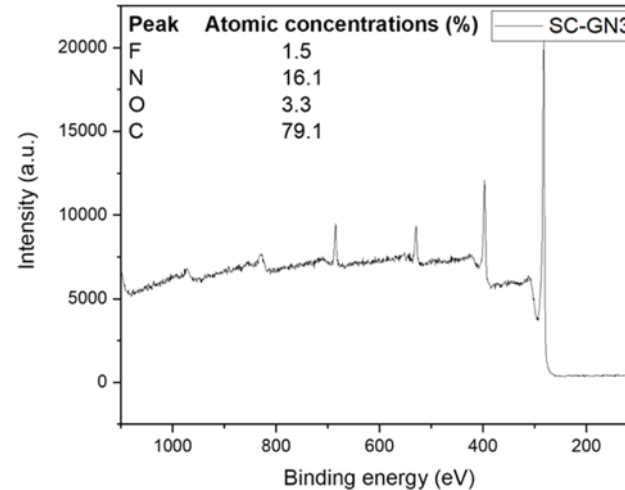
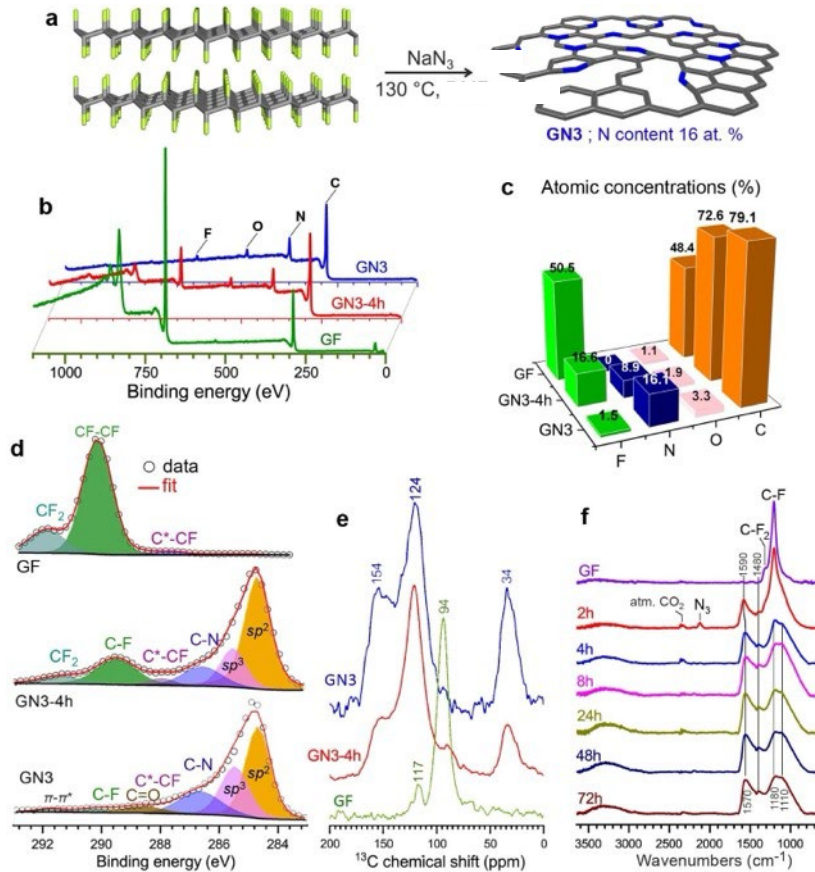
Energy storage mechanism is a physical process of ion accumulation on electrode material + electrolyte ion separation

Quick and reversible charging/discharging

Applications: requiring many rapid charge/discharge cycles (circuit protection, combined with batteries for recuperation etc.)



SC-GN3 synthesis via chemistry of fluorographene



SC-GN3 powder and example of 50 g packing of purified product.

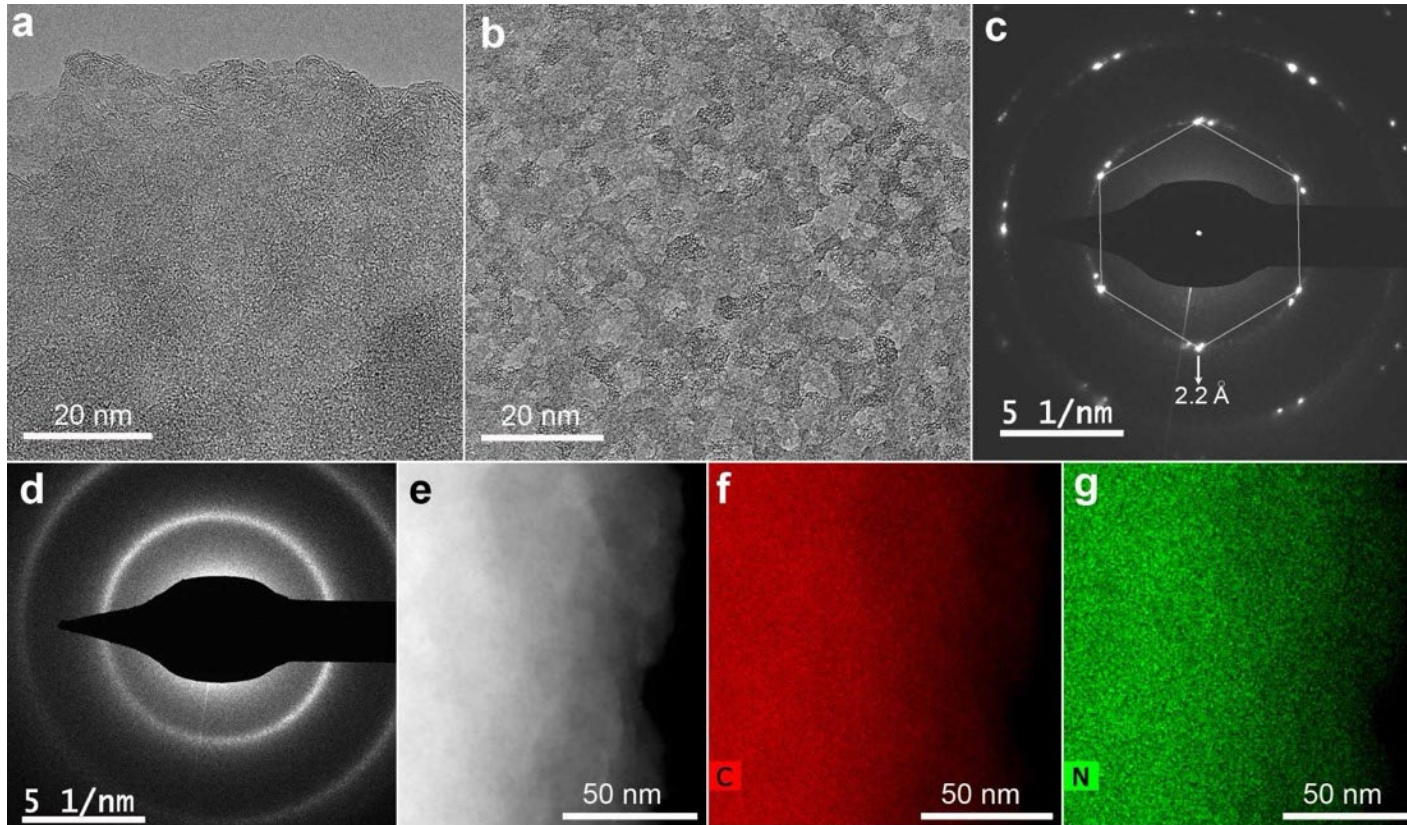
SC-GN3 material is highly N-doped (16 at. %) graphene-related material synthesized from graphite fluoride via wet chemistry in one step.

After synthesis purification steps are needed.

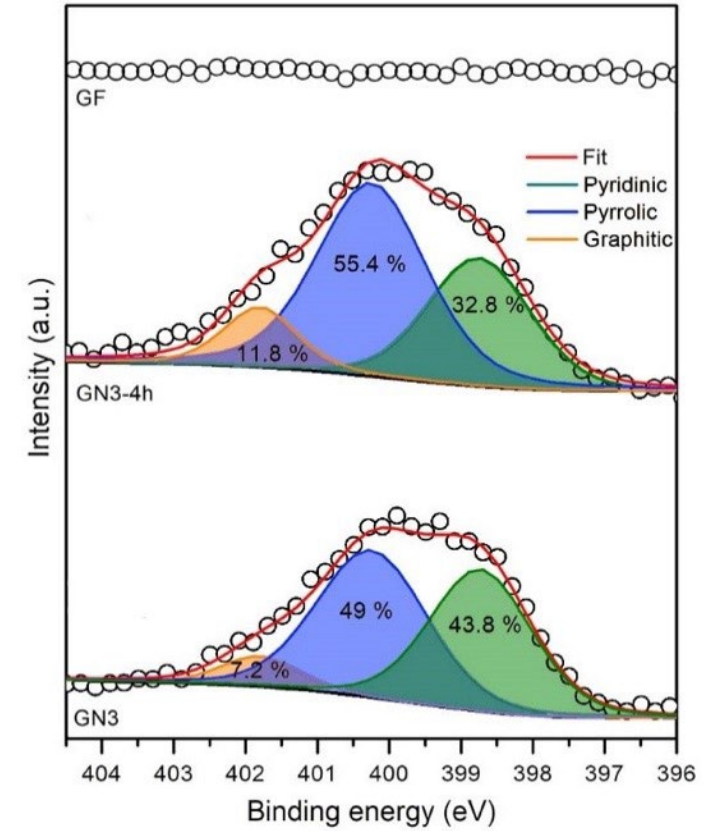


Reaction scheme, XPS, MS-NMR, and FTIR characterization of SC-GN3.

Characterization of SC-GN3

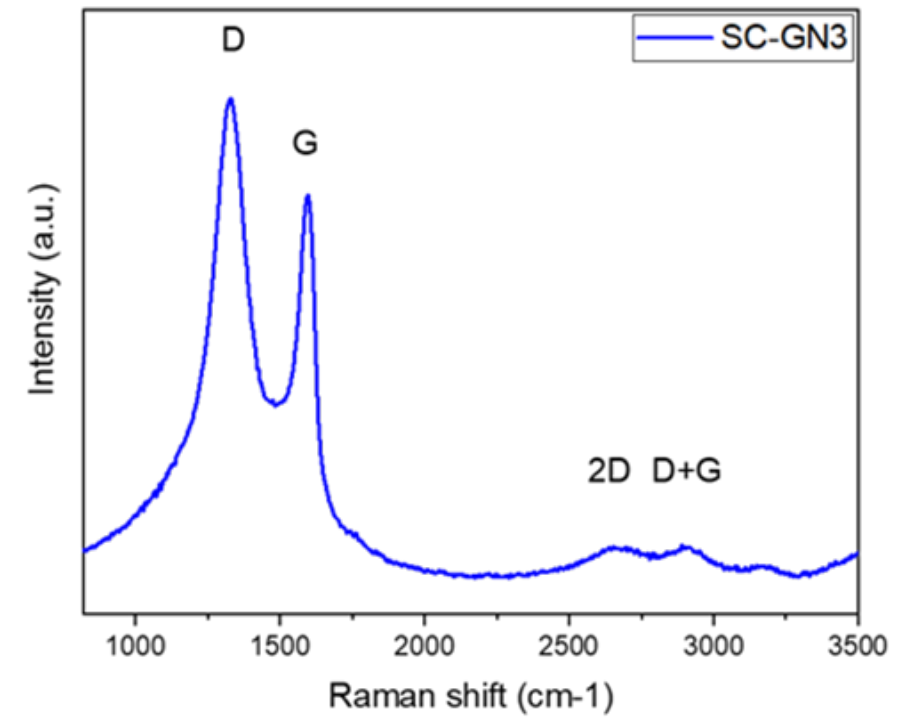
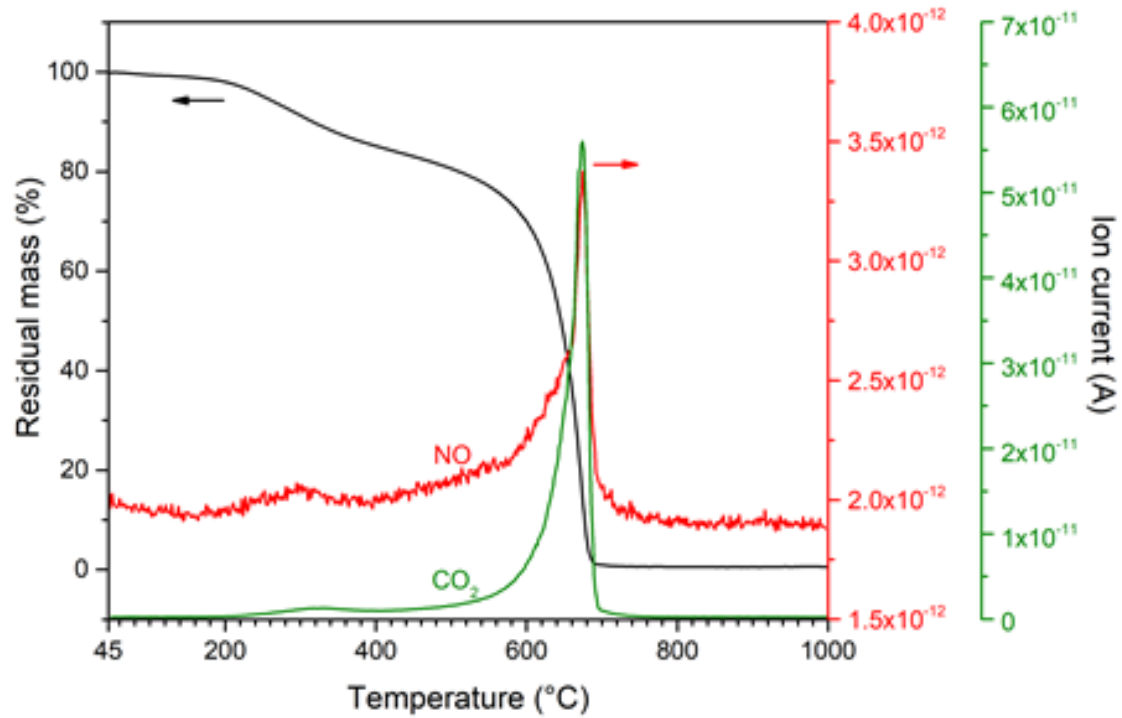


Homogeneous distribution of nitrogens in the lattice.



Chemical forms of nitrogens from HR XPS 1s N.

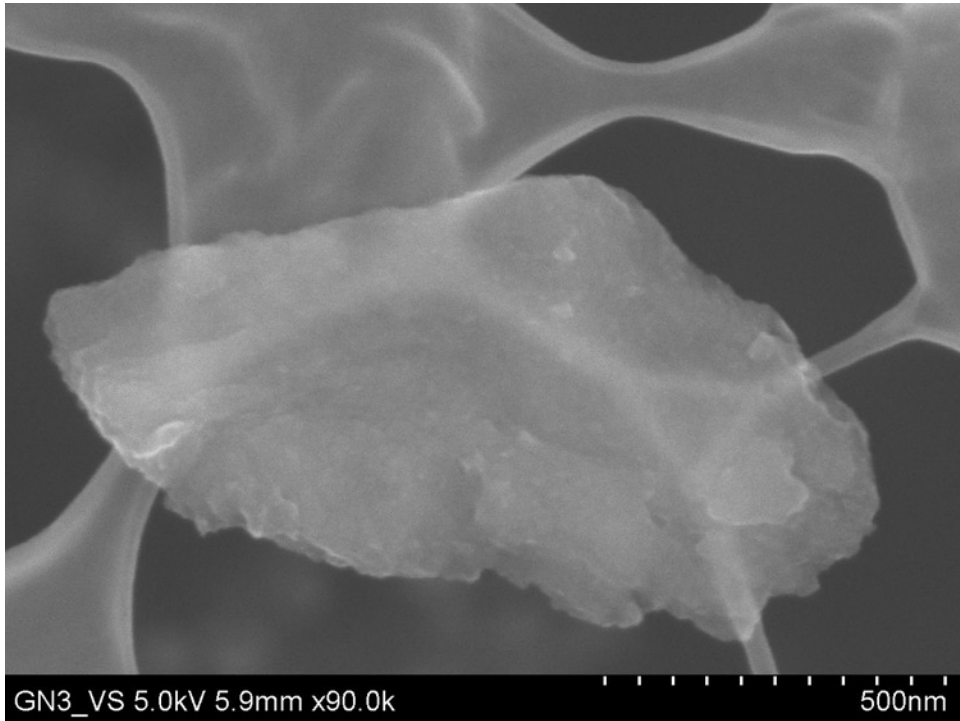
Characterization of SC-GN3



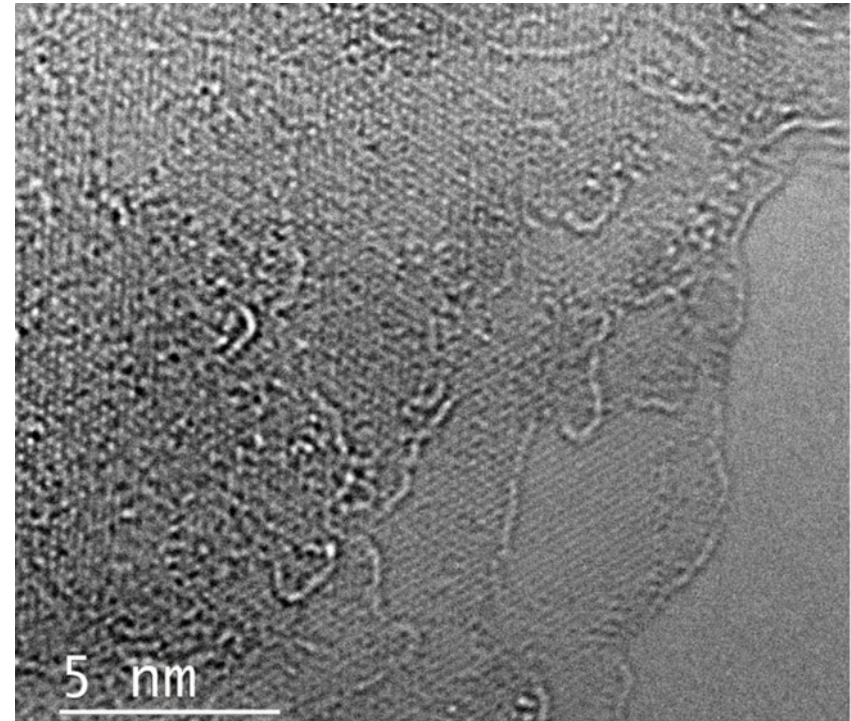
Thermal stability of SC-GN3 (with evolved gas analysis).

Raman spectrum of SC-GN3.

Characterization of SC-GN3

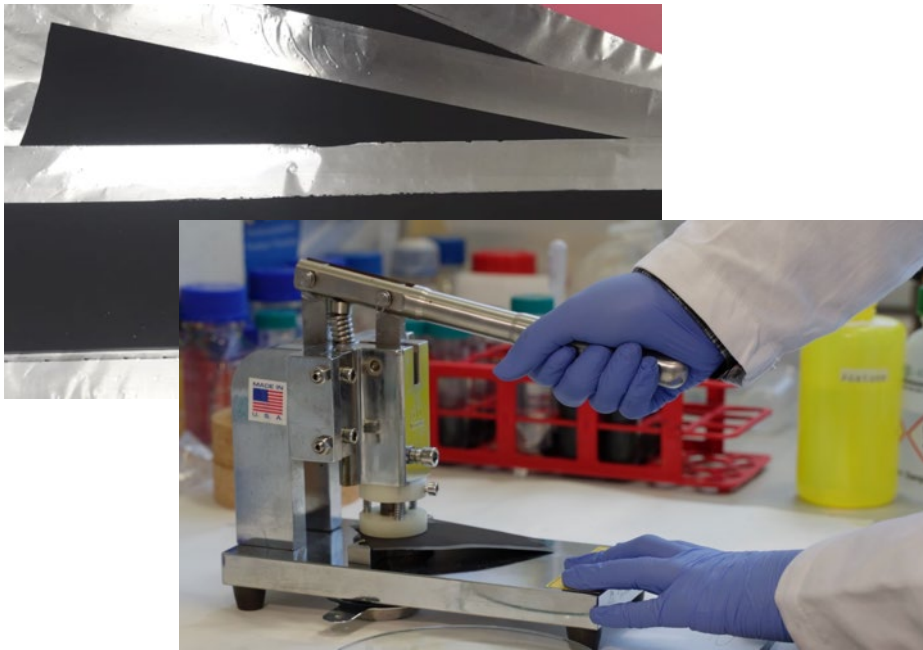


SEM image of N-doped graphene indicates on a few-layer structure.



TEM image of N-doped graphene

SC-GN3 testing

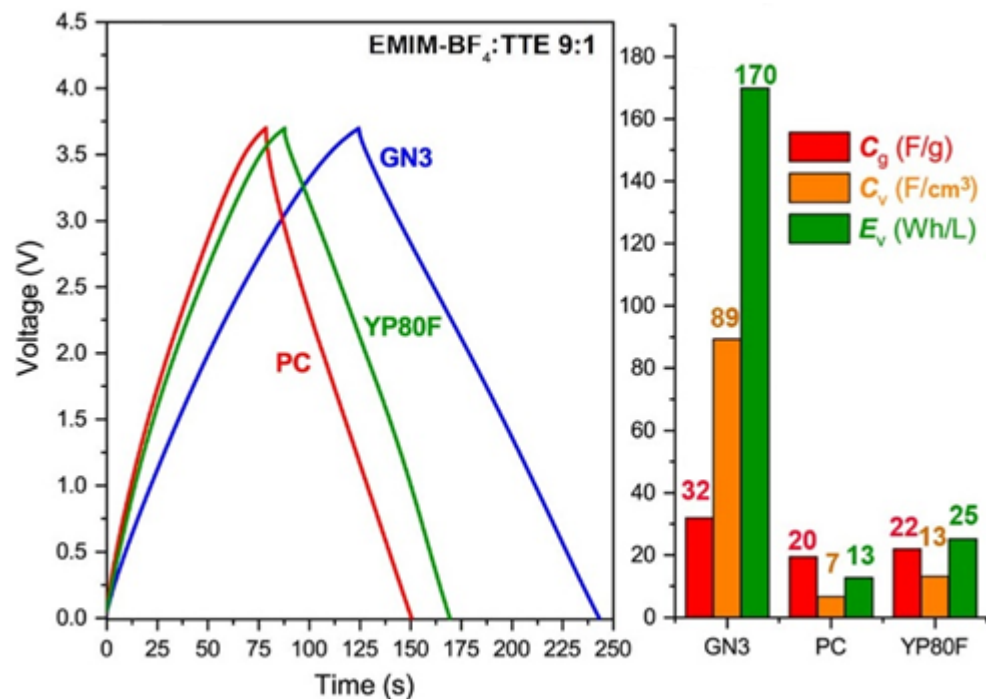
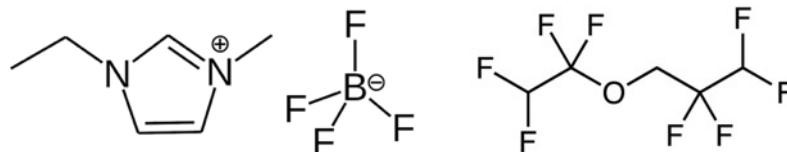


SC-GN3 is mixed with binder, electrodes are prepared.

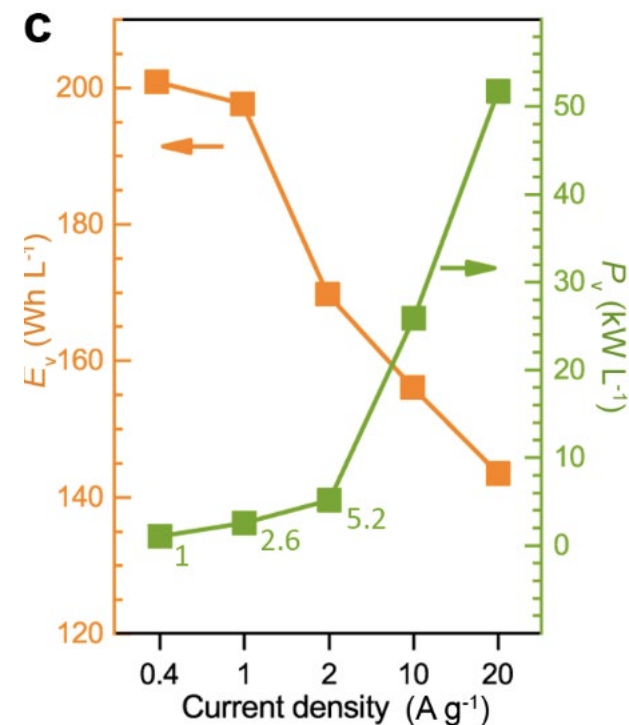


Scheme of EI-Cell used for testing (figures taken from el-cell.com).

SC-GN3

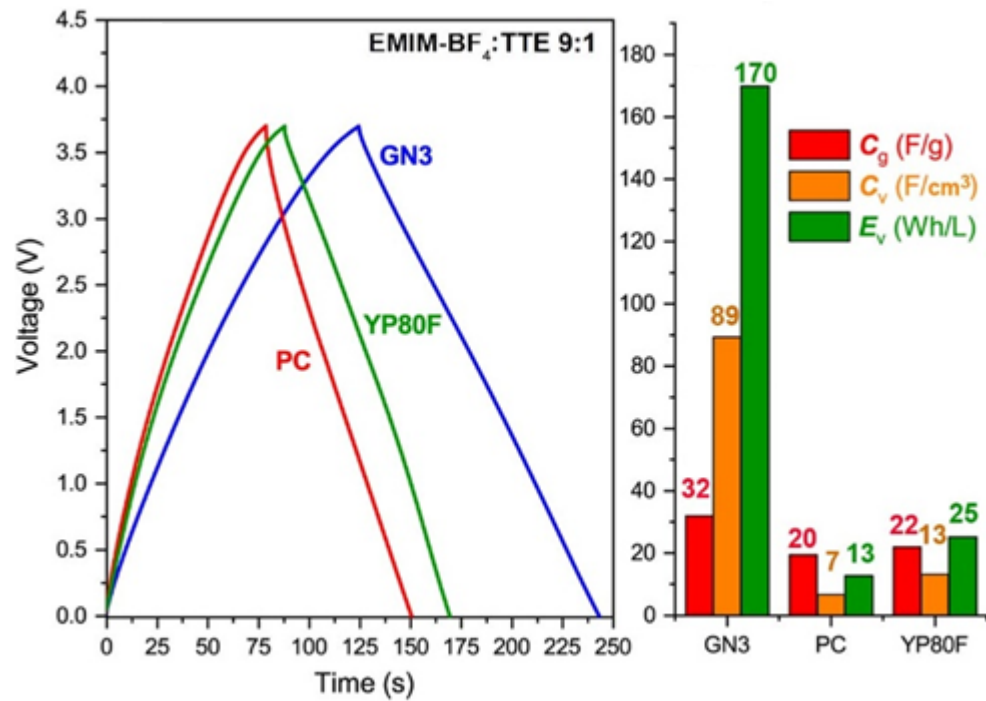


GCD profiles and supercapacitor performance comparison of commercial porous high surface area carbon materials and N-doped graphene (at 2 A g⁻¹).

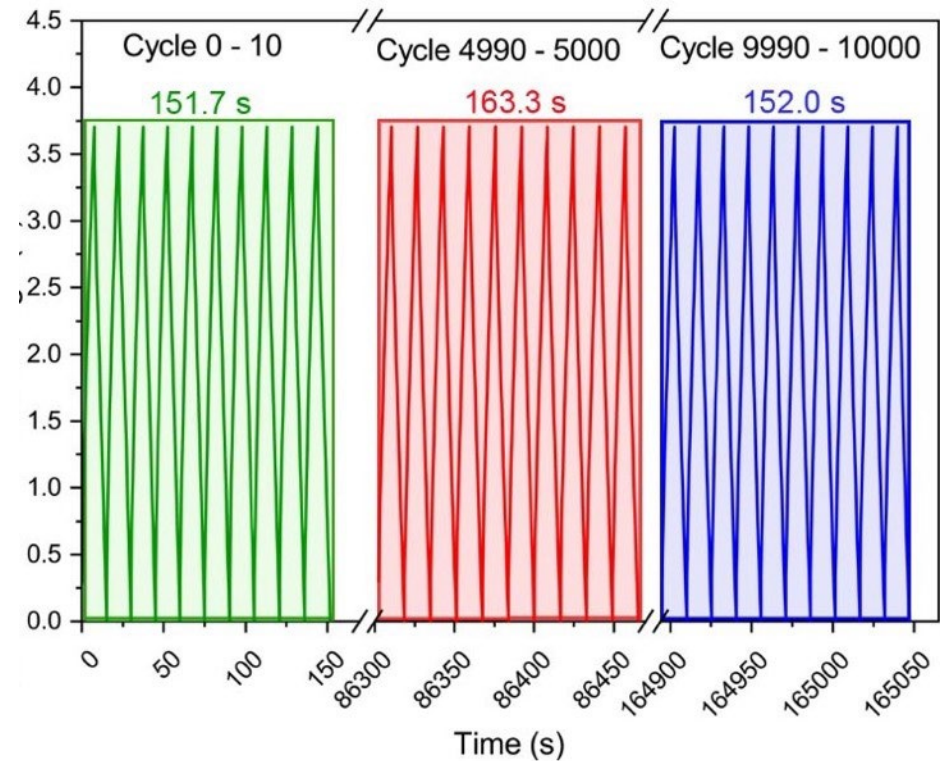


Energy and power density of SC-GN3 at increasing specific currents. SC-GN3 delivers energy densities of 200 Wh L⁻¹ at a power of 2.6 kW L⁻¹ and 143 Wh L⁻¹ at 52 kW L⁻¹.

SC-GN3

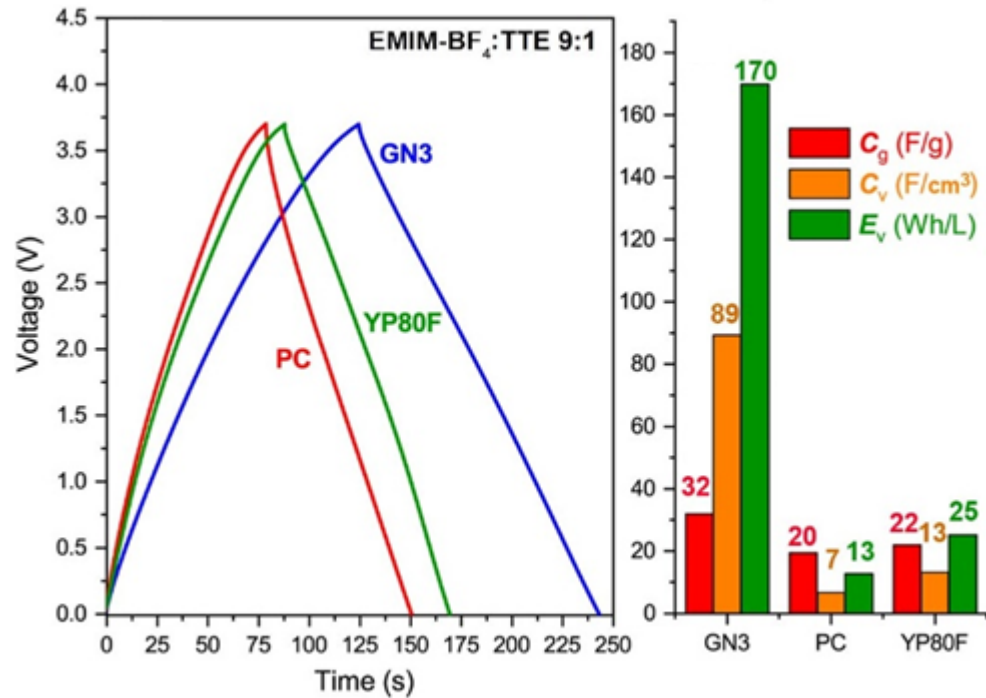


GCD profiles and supercapacitor performance comparison of commercial porous carbon materials and N-doped graphene.

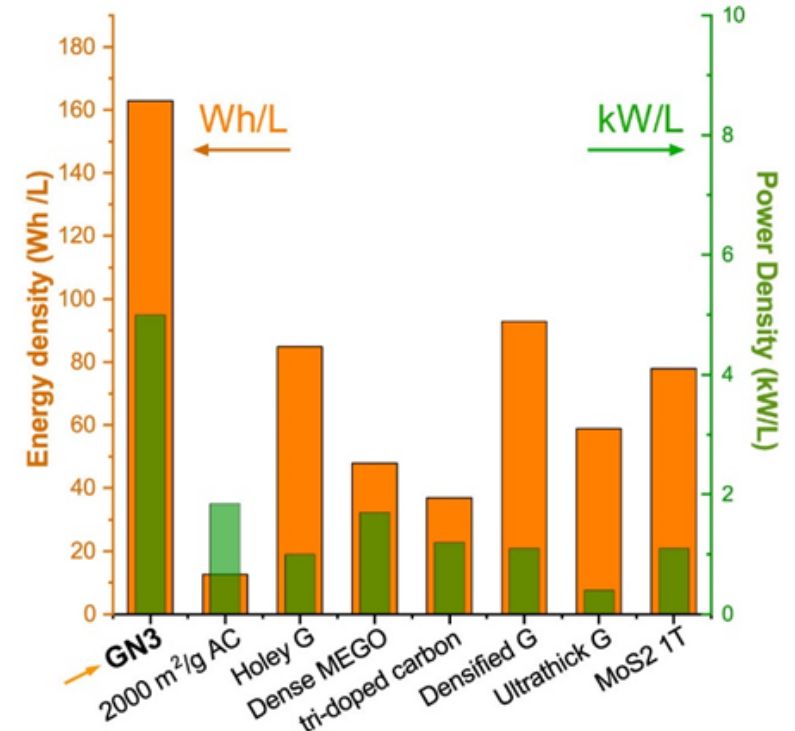


Stability of GN3 showing the GCD profiles at the beginning, mid-point, and end of a 10,000 cycle test.

SC-GN3



GCD profiles and supercapacitor performance comparison of commercial porous carbon materials and N-doped graphene.

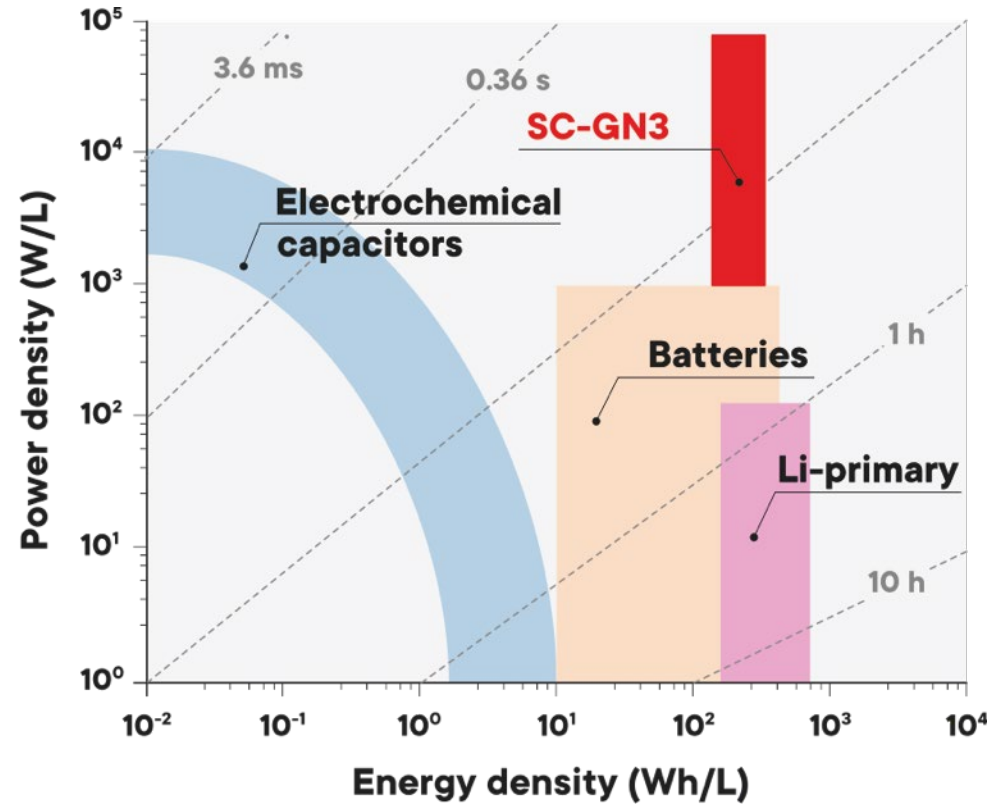


Current state-of-art comparison.

1. *Angewandte Chemie International Edition* 58, 2397–2401 (2019).
2. *Nature Communications* 5, 5554 (2014).
3. *Nano Energy* 2, 764–768 (2013).

4. *Science* 341, 534–537 (2013).
5. *Energy Environ. Sci.* 9, 3135–3142 (2016).
6. *Nature Nanotechnology* 10, 313–318 (2015)

SC-GN3



new material with record

energy density: 200 Wh/L
power density: 50 kW/L

Market Opportunity
Mini/Micro e-mobility

	BATTERIE	ENERGY-C
CONSTRUCTION	2 x 12V 75 Ah in series	6 x 5000F in series
RATED VOLTAGE	24V	24V
EFFECTIVE STORAGE ENERGY	1.800Wh	40Wh
RANGE	6 - 8h	700 meters (ca. 12 min)
CHARGE TIME	ca. 4h	<2min
VOLUME	16l	5l
WEIGHT	53kg	4,4kg (in future 2kg)
NUMBER OF CYCLES	~1000 cycles	>500.000 cycles

Prototypes in 2024

2025



Example of available devices

AVX PrizmaCap (4 Wh/L), rel. 7/21

Skeleton SkelCap (16 Wh/L), rel. 9/21

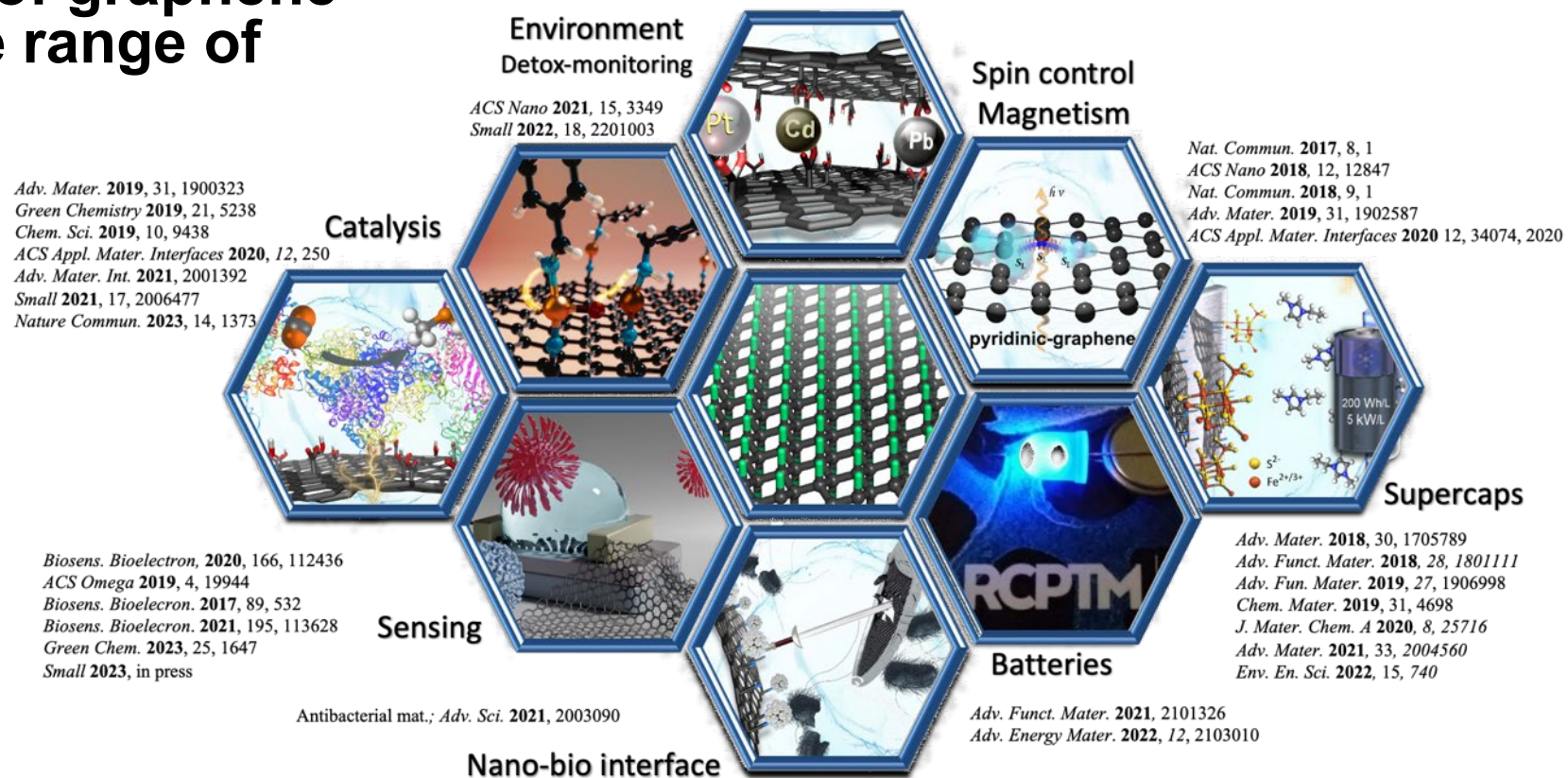
Driverless transport AGV
Automated Guided Vehicles



Source: Jianghai-Europe

Fluorographene

perspective material for
scalable synthesis of graphene
derivatives for wide range of
applications.



Acknowledgement

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Iosif Tantis (now Cornell)

Yevgen Obratsov

Vojtěch Kupka

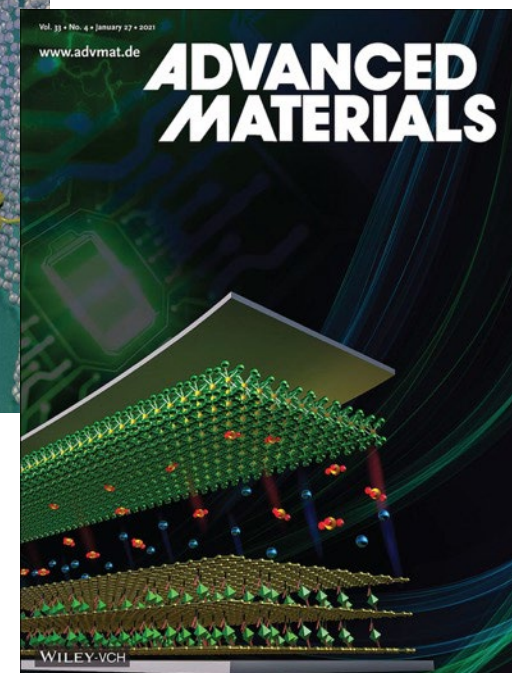
Miroslav Medveď

Vítězslav Hrubý

Veronika Šedajová (now Cambridge)

Tomáš Zedníček

Luca Primavesi



Graphene acid available at <https://graphene-derivatives.com/>

